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CHAPTER 12 REVIEW Solutions SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Match the type of mixture on the left to its representative particle diameter on the right. c solutions (a) larger than 1000 nm a suspensions (b) 1 nm to 1000 nm b colloids (c) smaller than 1 nm 2. Identify the solvent in each of the following examples:

## **mc06se cFMsr i-vi - nebula.wsimg.com**

CHAPTER 12 REVIEW Solutions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Solid CaCl<sub>2</sub> does not conduct electricity. Explain why it is considered to be an electrolyte. 2. Explain the following statements at the molecular level: a. Generally, a polar liquid and a nonpolar liquid are immiscible. b.

## **mc06se cFMsr i-vi - Ed W. Clark High School**

CHAPTER . 12 . REVIEW ""-- Solutions . SHORT ANSWER Answer

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the following questions in the space provided. 1. Describe the errors made by the following students in making molar solutions. 3. James needs a 0.600 M solution of KCl. He measures out 0.600 g of KCl and adds 1 L of water to the solid. James made several errors.

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solutions, but thicker and are usually not transparent, lastly ...

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Modern Chemistry 109 Solutions CHAPTER 12 REVIEW Solutions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Solid CaCl<sub>2</sub> does not conduct electricity. Explain why it is considered to be an electrolyte. \_\_\_\_\_

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Modern Chemistry Chapter 12-. Solutions. Section 1- Types of Mixtures Solutions are homogeneous mixtures of two or more substances in a single phase. Soluble describes a substance as capable of being dissolved. Solvent is the dissolving medium in a solution.

## **Modern Chemistry Chapter 12- Solutions**

42 42 Discussion Activity: Chapter 12 Research Assignment Review the following research activity, and post a response given the instructions below. In recent years, the United States Postal Service (USPS) has experienced declining revenues, largely due to the increase in digital communications such as e-mail.

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